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## Chapter 02 Testbank

Student: \_\_\_\_\_

- Anything that occupies space and has mass is called
  - an electron.
  - living.
  - matter.
  - energy.
  - space.
- The electrons of an atom are
  - always equal to the number of neutrons in an atom.
  - found in the nucleus.
  - used to determine atomic number.
  - positively charged.
  - moving in pathways called orbitals.
- All of the following pertain to the atom  $^{14}\text{C}$  except (E)
  - has 6 protons.
  - has 6 electrons.
  - has 14 neutrons.
  - is an isotope of carbon.
  - mass number is 14.
- The subatomic particles that surround the nucleus are the
  - electrons.
  - protons.
  - neutrons.
  - protons and neutrons.
  - protons and electrons.
- Cations are
  - charged subatomic particles.
  - atoms that have gained electrons.
  - radioactive isotopes.
  - capable of forming ionic bonds with anions.
  - atoms without protons.

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