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Cool! I'am really happy

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#Diego Butler



so many fake sites. this is the first one which worked! Many thanks

POST LAB QUESTIONS

3.) Consider the following equilibrium in a closed vessel:

$$\text{CaO}_{(s)} + \text{CO}_{2(g)} \rightleftharpoons \text{CaCO}_{3(s)}$$

a.) some CaO is added to the equilibrium mixture
b.) some CaCO₃ is removed
c.) some CO₂ is added
d.) a few drops of HCl solution is added
e.) temperature is lowered

a.) forward
b.) forward
c.) forward
d.) backward
e.) BONUS

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